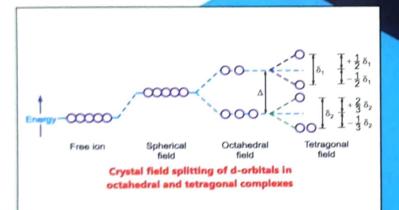
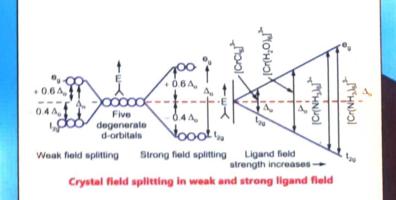
As Per New Syllabus (CBCS Pattern and NEP 2020)
Shivaji University, Kolhapur
B.Sc. Part-III: Semester-V

INORGANIC CHEMISTRY

CHEMISTRY (DSE-E5): PAPER-IX

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Prof. (Dr.) REKHA A. NALAWADE Prof. (Dr.) PRAVINA B. PISTE
Prof. (Dr.) ARJUN S. KUMBHAR









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As Per New Revised CBCS and NEP-1.0 Pattern Syllabus of Shivaji University, Kolhapur w.e.f. June 2024

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Preface

This book is basically intended for B.Sc. Part-III students for jemester V Course of Shivaji University, Kolhapur. This book is written according to new NEP-1.0 Pattern syllabus being implemented from June 2024 prescribed by Shivaji University, Kolhapur.

It is our great pleasure to present this book to the students and espected teachers in proper time. The subject matter is presented in imple and lucid language. This book covers all the chapters mentioned n the syllabus. The material is presented in a comprehensive way and the equence of articles in each chapter helps the students to understand the subject with ease.

Different diagrams and illustrative description is provided to enhance he learning and understanding of the matter by students as well as to enable the teachers to explain the difficult concepts properly. The solved numerical examples, long answer type questions, short answer type questions including multiple type questions are given at the end of each thapter.

We are thankful to Nirali Prakashan, Pune for making us a part of heir team of Authors. We thank **Mr. Dineshbhai Furia** and **Mr. Jignesh Furia** for publishing this book.

Last but not the least we are very much indebted to Mr. Virdhaval Shinde (Marketing Executive, Kolhapur District) and Mr. Ashok Nanavare (Marketing Executive, Sangli District) for their nice co-operation. We are very much thankful to Mr. Kiran Kamble (Proof Reading), Mrs. Anjali Muley (Graphic Design) and Mr. Malik Shaikh for a neat and error free D.T.P. of this book.

We hope that this book will be found useful by students and teachers. We will appreciate any suggestions for the improvement of the book.

Syllabus

Ur	it 1	: Acids	s, I	Bases	and Non	ı-aqueou	s Solvents	[06]
1								concent

- Introduction to theories of acids and bases Arrhenius concept, Bronsted-Lowry concept, Lewis concept, Lux-Flood concept (definition and examples)
- 1.2 Hard and soft acids and bases (HSAB concept)
 - 1.2.1 Classification of acids and bases as hard, soft and border line.
 - 1.2.2 Pearson's HSAB concept.
 - 1.2.3 Acid-Base strength and hardness softness.
 - 1.2.4 Applications and limitations of HSAB principle.
- 1.3 Chemistry of non-aqueous solvents
 - 1.3.1 Introduction, definition and characteristics of solvents
 - 1.3.2 Classification of solvents
 - 1.3.3 Physical properties and acid-base reactions in liquid ammonia (NH₃) and liquid sulphur dioxide (SO₂).

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- 2.1 Crystal Field Theory (CFT)
- 2.2 Introduction: Shapes of d-orbitals, Basic assumptions of CFT
 - 2.2.1 Crystal field splitting of d-orbitals of metal ion in octahedral, tetrahedral, square planar complexes and Jahn-Teller distortion.
 - 2.2.2 Factors affecting the crystal field splitting.
 - 2.2.3 High spin and low spin octahedral complexes w.r.t. Co(III).
 - 2.2.4 Crystal field stabilization energy (CFSE), Calculation with respect to octahedral complexes only.
 - 2.2.5 Limitations of CFT.
- 2.3 Molecular Orbital Theory (MOT)
 - 2.3.1 Introduction.
 - 2.3.2 MOT of octahedral complexes with sigma bonding such as $[Ti(H_2O)_6]^{3+}$, $[CoF_6]^{3-}$, $[Co(NH_3)_6]^{3+}$.
 - 2.3.3 Merits and demerits of MOT.

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